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A furnace shell has to be cooled from 90 °Cto 55 °C. The mass of the furnace shell is 2 tonnes; the specific heat of furnace shell is 0.2 kcal/kg°C. Water is available at 28 °C. The maximum allowed increase in water temperature is 5 °C. Calculate the quantity of water required to cool the furnace. Neglect heat loss. [2800 kg]



Examples	
Production rate from a paper machine is 340 tonnes per day (TPD). Inlet and outlet dryness to paper machine is 40% and 95% respectively. Evaporated moisture temperature is 80 °C. To evaporate moisture, the steam is supplied at 35 kg/cm². Latent heat of steam at 35 kg/cm² is 513 kcal/kg. Assume 24 hours/day operation a) Estimate the quantity of moisture to be evaporated b) Input steam quantity required for evaporation (per hour). Consider enthalpy of evaporated moisture as 632 kCal/kg. [24 TPH]	
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