

# ME 6101: Classical Thermodynamics

Term Paper 02

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1. A heat engine that receives heat from a furnace  $1200^{\circ}\text{C}$  and rejects waste heat to a river at  $20^{\circ}\text{C}$  has a thermal efficiency of 40 percent. Determine the second-law efficiency of this power plant.
2. A house that is losing heat at a rate of  $50,000\text{ kJ/h}$  when the outside temperature drops to  $4^{\circ}\text{C}$  is to be heated by electric resistance heaters. If the house is to be maintained at  $25^{\circ}\text{C}$  at all times, determine the reversible work input for this process and the irreversibility.
3. A turbine receives steam at  $3000\text{ kPa}$ ,  $500^{\circ}\text{C}$  and has two exit flows, one at  $1000\text{ kPa}$ ,  $350^{\circ}\text{C}$  with 20% of the flow and the remainder at  $200\text{ kPa}$ ,  $200^{\circ}\text{C}$ . Find the isentropic and second-law efficiencies.
4. Steam enters a turbine at  $25\text{ MPa}$ ,  $550^{\circ}\text{C}$  and exits at  $5\text{ MPa}$ ,  $325^{\circ}\text{C}$  at a flow-rate of  $70\text{ kg/s}$ . Determine the total power output of the turbine, its isentropic efficiency, and its second-law efficiency.
5. Air enters a compressor at ambient conditions of  $100\text{ kPa}$ ,  $300\text{ K}$  and exits at  $800\text{ kPa}$ . If the isentropic compressor efficiency is 85%, what is the second-law efficiency of the compressor process?
6. A compressor is used to bring saturated water vapor at  $1\text{ MPa}$  up to  $15\text{ MPa}$ , where the actual exit temperature is  $650^{\circ}\text{C}$ . Find the irreversibility and the second-law efficiency.
7. A heat exchanger brings  $10\text{ kg/s}$  water from  $100^{\circ}\text{C}$  up to  $500^{\circ}\text{C}$  at  $2000\text{ kPa}$  using air coming in at  $1400\text{ K}$  and leaving at  $460\text{ K}$ . What is the second-law efficiency?
8. Saturated R-134a vapour enters the adiabatic compressor of vapour compression refrigeration cycle at  $0.2\text{ MPa}$  and leaves the condenser as a saturated liquid at  $0.8\text{ MPa}$ . The temperature of the vapor leaving the irreversible compressor is  $44^{\circ}\text{C}$  and the ambient temperature is  $25^{\circ}\text{C}$ . Determine the energy and availability changes and irreversibilities around the cycle.

9. Refrigerant-134a at 1 MPa and 100°C is throttled to a pressure of 0.8 MPa. Determine the reversible work and exergy destroyed during this throttling process. Assume the surroundings to be at 30°C.
  
10. Steam enters an adiabatic nozzle at 3.5 MPa and 300°C with a low velocity and leaves at 1.6 MPa and 250°C at a rate of 0.4 kg/s. If the ambient state is 100 kPa and 18°C, determine (a) the exit velocity, (b) the rate of exergy destruction, and (c) the second-law efficiency.